

## Physicochemical quality analyses of Camembert cheese packed in modified atmosphere using perforated film

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### ABSTRACT

*Camembert is a mould-ripened soft cheese that is highly perishable because of its high nutritional value and moisture content. This study examined the effect of using two different films of different permeability on the physicochemical quality of the cheese samples. The samples were prepared at Cheese cellar, London, and analysed at the science laboratory of London South Bank University, London. The two films used were of the same food grade material i.e. polyethylene terephthalate (PET). One film had perforations (Mp), while the other is a non-perforated film (M). Without perforations, the permeability of the film for water, O<sub>2</sub>, CO<sub>2</sub> and N<sub>2</sub> is 3.9g/m<sup>2</sup>/24hrs at 38<sup>o</sup>C, 3cm<sup>3</sup>/m<sup>2</sup>/24hrs at 23<sup>o</sup>C, 30cm<sup>3</sup>/m<sup>2</sup>/24hrs at 23<sup>o</sup>C and 3cm<sup>3</sup>/m<sup>2</sup>/24hrs at 23<sup>o</sup>C respectively.*

*The two set of samples were stored at 4<sup>o</sup>C ± 0.5 and studied for a period of three weeks. The samples were randomly selected in duplicates on day 0 (of week 1), day 7 (of week 2), and day 14 (of week 3). Changes in the physicochemical properties of the food samples were studied.*

*The effect of the different films on intrinsic parameters such as the pH, water activity (a<sub>w</sub>) and moisture content (Mc) of the food sample was measured.*

*For sample Mp, there was a steady increase in the pH, while sample M had an initial decrease in pH, followed by a rise in pH. Generally, increase in pH for sample Mp was higher than that of sample M. For both sample groups, there was a steady decrease in a<sub>w</sub>, while there was an initial decrease in Mc followed by an increase.*

*Based primarily on physicochemical evaluation, the use of the perforated film (Mp) has a good potential for preserving the keeping quality of the food sample.*

### 1. INTRODUCTION AND LITERATURE REVIEW

Camembert is an important mould-ripened soft cheese which is commonly consumed in many countries for its peculiar taste and nutritional value. An important determinant of the shelf-life of cheeses is their moisture content. It has been noted that cheeses having high pH and moisture content, such as Camembert are very susceptible to microbial spoilage, by moulds, yeasts, and enteric bacteria. (Papaioannou et al, 2007).

Therefore, the extension of the shelf life of this soft cheese, and keeping it fresh for a longer period of time is a matter of importance.

Consumer demands for natural, healthy, preservatives-free products with extended shelf-life, have led food technologists and researchers to develop new packaging concepts. Among these new packaging concepts is the modified atmosphere packaging (MAP). MAP alters the natural gas composition surrounding the product in the package in order to delay deteriorative changes (Air products, 1995; Maria R. B., 2009). MAP technologies increase commercial life of cheeses

because they combine the protection against oxidation and dehydration with the inhibition of undesirable microorganisms (Olivares *et al*, 2012)

MAP of different cheese varieties has been studied in the last decades (Rodriguez-Aguilera *et al*, 2011), but scarce information related to MAP of Camembert cheese is available.

The permeability and the transmission rate of the packaging film to O<sub>2</sub>, CO<sub>2</sub>, and water vapour are among the most essential factors which determine the gas composition in the package, which may

For packaged cheeses, the thermodynamic driving force for water transfer out of the cheese and out of the package depends also on the barrier to moisture that the package offers (Holm, Mortensen, & Risbo, 2006).

The cheese under study is a cylindrical, cream coloured, mould-ripened cheese. Traditionally, Camembert tends to be sold whole in thin, round, wooden containers, but it is also sold in tins, with a ring-pull tab for opening. The product is also commonly wrapped dry in a paper/foil wrapper. These systems of packaging may lead to excessive water loss from the product. On the other hand, using a barrier film that has very low permeability to water will not produce a desirable quality for these cheeses. The Camembert cheese under study does not last more than seven days after it has been unwrapped, cut and repacked by the retailer in a MAP, using non-perforated film.

The shelf life of the Camembert cheese product used in this study could further be reduced to three days after unwrapping and cutting it.

The shelf-life of this type of cheese could be extended by allowing water loss, but at a lower rate. Selecting a packaging film with the right permeability to water could be challenging, but

influence the product's deterioration rate (Mullan and McDowell 2003; Church 1994). Therefore, the MAP design for a product requires careful handpicking of the packaging film type and size of packaging for the product (Farber *et al*. 2003). The changes in water activity (a<sub>w</sub>) and moisture content (Mc) that accompanies the maturation and storage of cheeses influences their microbiological and physicochemical properties (Saurel, Pajonk, & Andrieu, 2004). Therefore, the control of a<sub>w</sub> and moisture content is very important for the preservation of quality and safety of cheeses.

Identifying the correct packaging film for cheeses has been noted to be the key to extending the shelf-life of this type of cheeses (Simal, Sa'nchez, Bon, Femenia, & Rossello', 2001). The use of a packaging system with a tailor-made moisture barrier, which allows for water loss, but at a lower rate, is a way of extending the cheese's shelf-life (Pantalea~o I. *et al*., 2006). MAP applications for a product are developed from different trials and tests of different methods (Air product, 1995). The final choice on the packaging film to be used should be made based on vast evaluations and in collaboration with the packaging suppliers (Sandhya, 2010). Although, MAP with 30% CO<sub>2</sub> and 70% N<sub>2</sub> is recommended for soft cheeses, it is not used for mould-ripened soft cheeses (Air product, 1995). Also, perforated films are often used for fresh produce such as fruits and vegetables i.e. this is the first study evaluating the effect of MA and a perforated film on Camembert cheese.

The study examined the effect of using two films of same food grade material (PET), but of different permeability, and a modified atmosphere of 60% CO<sub>2</sub> and 40% N<sub>2</sub> on the sensory and physicochemical quality of Camembert cheese.

## 2. MATERIALS AND METHOD

The methodology used for the study is described below.

### 2.1 Food sample preparation

The samples were collected from the delivery unit of *Cheese cellar*, London, and taken to the production unit (of the same food company), where each was unwrapped and cut into 8 wedges, using a clean wedge line cutter. The wedges were placed in trays (each wedge weighing approximately 30g).

The packaging trays were then separated into 2 groups.

The Multivac's MAP machine which uses preformed tray and lidding film (PTLF) was used for the sample preparation. The Multivac's MAP machine was used to evacuate air from the package and to seal the trays with the 2 different films, after the addition of protective gases (CO<sub>2</sub>, N<sub>2</sub>) at pre-set levels. The gas mixture used for the study was 60% CO<sub>2</sub> and 40% N<sub>2</sub>.

The samples were then stored at 4°C ± 0.5 in the storage room of the factory. Duplicate samples were randomly collected and analysed. The samples were held at ambient temperature during the evaluation of the physicochemical quality of the samples. A two weeks trial study was conducted, before the final three weeks study. The different samples were labelled using different code. Letter *M* was used for the camembert cheese packaged using non-perforated film, while letter *M<sub>p</sub>* was used for the camembert cheese packaged using perforated film. Numeric subscripts such as 1, 2,

and 3 were used to differentiate samples collected in different weeks (i.e. 1 for day 0 of week 1, 2 for day 7 of week 2, and 3 for day 14 of week 3)

### 2.2 Physicochemical analyses

#### (a) Package headspace gas composition:

Systech's gaspace advance headspace gas analyser was used to analyse the package headspace gas composition. The percentage CO<sub>2</sub>, N<sub>2</sub> and O<sub>2</sub> in the two sets of packages was monitored weekly.

The result, reported as mean value, was used to compare the permeability of the two films to the different gases.

#### (b) pH measurement

The pH value of each sample was recorded using a digital pH meter (Mettler Toledo FE20 Desktop pH Meter), equipped with a glass electrode that was inserted directly into the cheese sample for the measurement.

#### (c) Water activity and Moisture content measurement

The water activity value for the samples was determined using Thermoconstanter novasina TH200 water activity (aw) meter, while the moisture content value for the samples was determined using the AnD ML-50 Moisture Analyser.

The water activity values for each sample were determined at predetermined temperature for 5 min, while the moisture content values for each sample were determined by weight loss via heating at 105 °C for 10 min.

### 3. RESULTS AND DISCUSSION

#### 3.1 Physicochemical analyses of the two set of samples

Changes in the physicochemical attributes (pH, water activity, moisture content) were monitored during storage of the two set of samples.

The headspace gas composition was also measured throughout the period of study. The results are shown below.

##### (a) pH properties

<i>Preliminary study</i>		
Week	Ma pH	Mpa pH
1	6.05	6.04
2	5.65	6.47
<i>Final study</i>		
Week	Mb pH	Mpb pH
1	5.92	5.83
2	5.65	6.71
3	6.07	6.87

Table 1. Mean pH of the Samples.

Ma = samples from preliminary study packed using non-perforated film. Mb= samples from final study packed using non-perforated

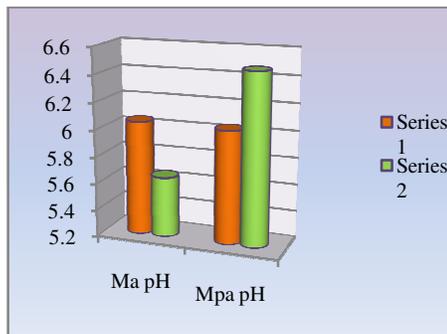


Fig . 1 Mean pH of sample Ma and Mpa

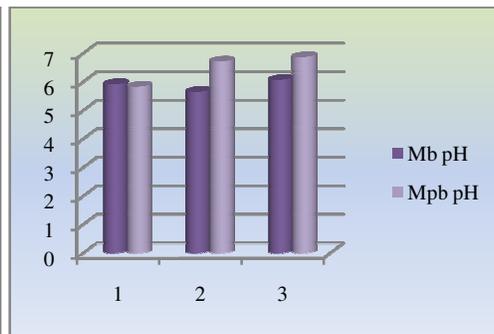


Fig . 2 Mean pH of sample Mb and Mpb

(b) Water activity (aw) measurements

<i>Preliminary study</i>		
Week	Ma aw	Mpa aw
1	0.615	0.547
2	0.558	0.514
<i>Final study</i>		
Week	Mb aw	Mpb aw
1	0.716	0.676
2	0.665	0.613
3	0.468	0.454

Table 2. Mean water activity of the samples. Ma = samples from preliminary study packed using non-perforated film, Mb= samples from final study packed

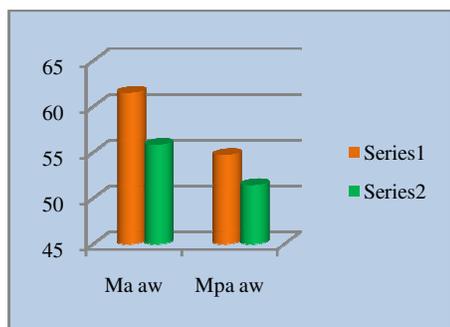


Fig. 3 Mean water activity of sample Ma and Mpa

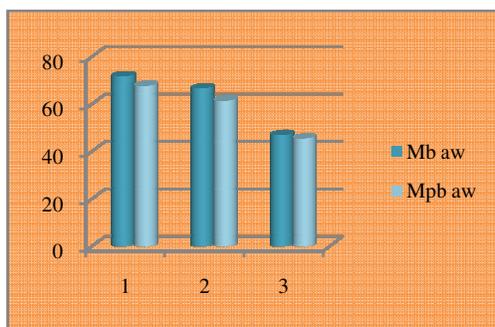


Fig. 4 Mean water activity of sample Mb and Mpb

(c) Moisture content(Mc) measurements

<i>Preliminary study</i>		
Week	Ma Mc	Mpa Mc
1	59.0	57.0
2	58.0	48.8
<i>Final study</i>		
Week	Mb Mc	Mpb Mc
1	59.0	47.0
2	58.0	43.0
3	98.0	77.0

Table 3. Mean moisture content of the samples. Ma = samples from preliminary study packed using non-perforated film, Mb= samples from final study packed using non-perforated film, Mpa =

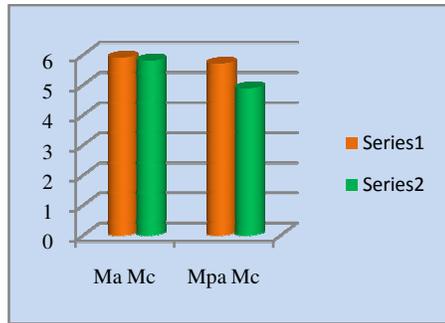


Fig. 5 Mean Moisture content of sample Ma and Mpa

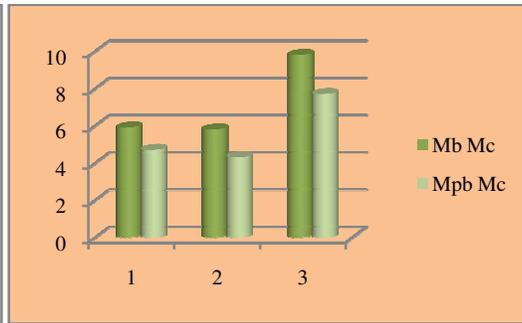


Fig. 6 Mean Moisture content of sample Ma and Mpa

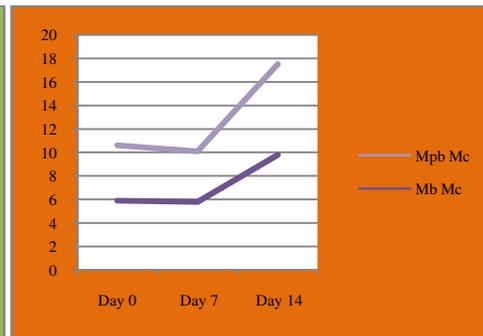
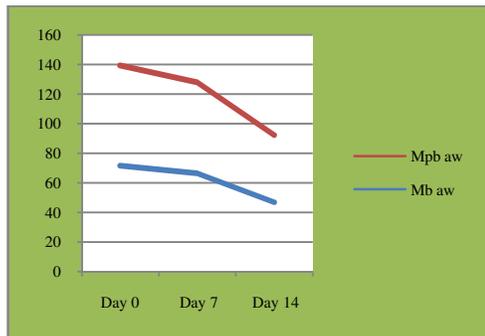


Fig. 7 Changes in water activity ( $a_w$ ) for both samples M and Mp Fig. 8 Changes in moisture content (Mc) for both samples M and Mp

(d) Mean headspace gas composition of modified atmosphere packaged (MAP) Camembert cheese using non-perforated (M) and perforated (Mp) film

First week	M1a Gas Composition	Mp1a Gas Composition
CO <sub>2</sub> %	62.4	1.3
N <sub>2</sub> %	34.1	77.8
O <sub>2</sub> %	3.59	20.9
Second week	M2a Gas Composition	Mp2a Gas Composition
CO <sub>2</sub> %	66.1	0.85
N <sub>2</sub> %	33.3	78
O <sub>2</sub> %	0.63	21.2

Table 4. Mean headspace gas composition of sample M and Mp for 1st and 2nd

First Week	M1b Gas Composition	Mp1b Gas Composition
CO <sub>2</sub> %	61.3	3.1
N <sub>2</sub> %	34.8	76
O <sub>2</sub> %	3.91	20.9
Second Week	M2b Gas Composition	Mp2b Gas Composition
CO <sub>2</sub> %	63.3	0.25
N <sub>2</sub> %	36.1	78.9
O <sub>2</sub> %	0.672	20.9
Third Week	M3b Gas Composition	Mp3b Gas Composition
CO <sub>2</sub> %	57	0.15
N <sub>2</sub> %	41.5	77.6
O <sub>2</sub> %	1.55	22.4

Table 5. Mean headspace gas composition of sample M and Mp for 1st, 2nd and

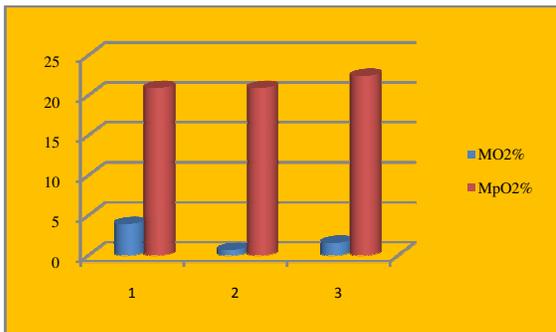


Fig. 9 Mean percentage Oxygen of sample M and Mp for 1st, 2nd and 3rd week of final study.

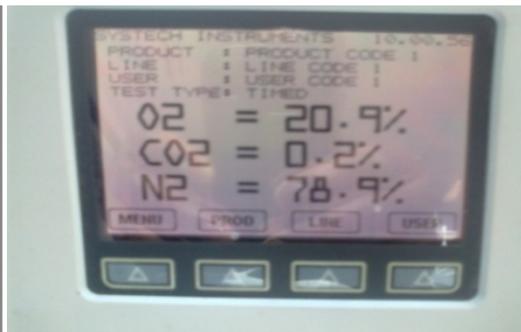


Fig. 10 Showing the typical reading of the gas analyser for the package sealed with perforated film, Mp.

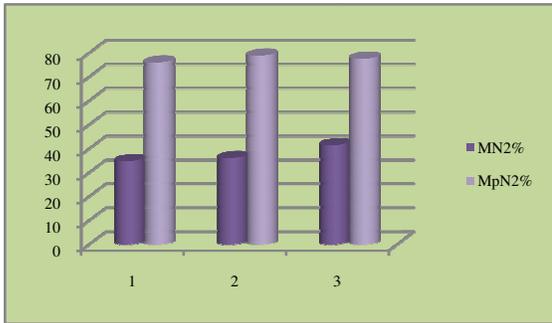


Fig. 11 Mean percentage Nitrogen of sample M and Mp for 1st, 2nd and 3rd week of final study.

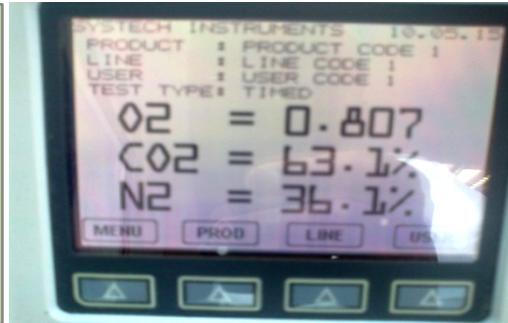


Fig. 12 Showing the typical reading of the gas analyser for the package sealed with non-perforated film, M.

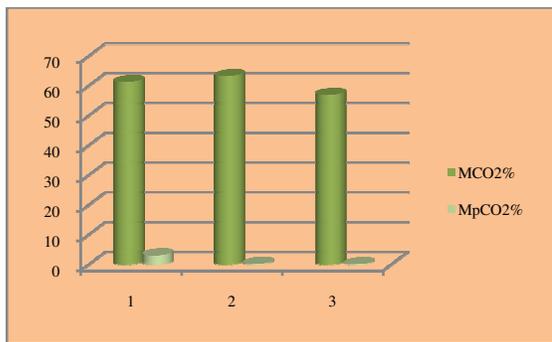


Fig. 13 Mean percentage Carbon dioxide of sample M and Mp for 1st, 2nd and 3rd week of final study.



Fig. 14 Showing water droplets condensed on the inner surface of the perforated film of sample Mp2.

(a) *Headspace gas composition:*

As expected, the headspace gas composition (CO<sub>2</sub>, N<sub>2</sub> and O<sub>2</sub>) for sample Mp was significantly different from the headspace gas composition for sample M, throughout the period of storage i.e. the perforated film was more permeable to the gases. The headspace gas composition for the package with perforated film was nearly that of the normal atmosphere by the third week of final study. See table 4 and 5. This difference in transmission rates is attributed to the perforations in the perforated film.

There was significant difference ( $P < 0.05$ ) between the headspace gas composition of the packages sealed with non-perforated film and the packages sealed with perforated film. The package sealed with perforated film ((Mp) allowed high amount of O<sub>2</sub> and N<sub>2</sub>, The package sealed with non-perforated film has low permeability. It therefore retained the high CO<sub>2</sub> concentration and

maintained the low amount of residual O<sub>2</sub>. Package Mp on the other hand allowed low amount of CO<sub>2</sub>. See fig. 9, 11, 13 and table 6, 7, and 8.

(b) *Changes in pH:*

For the cheese samples in trays sealed with perforated film (Mp), there was a steady increase in the pH (increase in alkalinity), while the cheese samples in trays sealed with non-perforated film (M) had an initial decrease in pH (increase in acidity), followed by a rise in pH. See fig. 1 and 2. Generally, the pH of sample Mp was higher than that of sample M, though the difference was not much. This steady increase in pH of sample Mp can be explained by considering the headspace CO<sub>2</sub> composition in the tray sealed with the perforated film. The % CO<sub>2</sub> in trays sealed with perforated film ranged from 0.15 to 3.1, while the %CO<sub>2</sub> in trays sealed with non-perforated film ranged from 57 to 66.1.

The high CO<sub>2</sub> concentration in sample M was expected to result in a pH drop, which is thought to be associated with the formation of carbonic acid, acidic amino acids, and free fatty acid production during proteolysis and lipolysis, respectively (Dermiki et al. 2008). Farber (1991) also reported pH reduction in food samples, caused by CO<sub>2</sub> atmosphere. The increase in the pH of sample Mp may be attributed to the utilization of lactic acid by *Penicillium* and the release of ammonia as an end product of its proteolytic and deaminating activities (Fox et al, 1993). The pH changes could also be explained by water transmission rates of the two films. The rate of water transmission for the perforated film is presumed to be more than that of the non-perforated film. This could have affected the pH. See fig. 14

(c) *Changes in Water activity*(a<sub>w</sub>):

Generally, there was a steady decrease in a<sub>w</sub> for both samples, though not significant to prevent bacterial and fungal growth. The decrease in water activity (a<sub>w</sub>) of both samples over the storage period, could be attributed to the proteolytic and lipolytic activities of the primary and secondary microflora of the cheese, which lead to production of several molecules e.g. amines, aldehydes, alcohols, sulphur compounds. Reduction in water activity has been attributed to proteolysis of casein (Schlesser et al, 1992). Also, the lowering of water activity during storage could be attributed to surface evaporation of water from the samples, through the films (Schlesser et al, 1992). See fig. 14

(d) *Changes in Moisture content*(Mc) :

The trend of change in Mc was same for both samples, though sample Mp had lower Mc throughout the study. There was a slight decrease in the Mc of both samples, between the first and second week of preliminary and final study.

Sample Mp had greater decrease in Mc, due to the perforations.

Water droplets about to escape can be seen to condense on the inner side of the perforated film in fig. 14. This could have resulted in the relatively lower moisture content of sample Mp. On the 14th day of the third week of final study, the Mc of sample M had increased appreciably, as compared to sample Mp. This could have contributed to the runny state of the camembert cheese in the tray sealed with the non-perforated film.

This increase in Mc may be due to solubilization of casein, an increase in pH, or both (Schlesser et al, 1992). Also, the mould responsible for ripening could have absorbed water for its metabolic activities.

### 3.2 Statistical analysis

The statistical analysis of the sets of data obtained was done using SPSS software (version 15). There was significant difference (P < 0.05) between the headspace gas composition of the packages sealed with non-perforated film and the packages sealed with perforated film at alpha level of 0.05 and confidence level of 0.95. The results of the statistical analyses as produced by the SPSS version 15 software are shown below.

Table 6. Statistical analysis (Independent Samples Test)for % O<sub>2</sub>

Barrier_film		N	Mean	Std. Deviation	Std. Error Mean
<b>Gas_Mix_O<sub>2</sub>%_Result</b>	<i>Non_Perforated_Film</i>	3	2.0440	1.67457	.96681
	<i>Perforated_Film</i>	3	21.4000	.86603	.50000

		Levene's Test for Equality of Variances		t-test for Equality of Means						
		F	Sig.	t	df	Sig. (2-tailed)	Mean Difference	Std. Error Difference	95% Confidence Interval of the Difference	
									Lower	Upper
<b>Gas_Mix_O<sub>2</sub>%_Result</b>	<i>Equal variances assumed</i>	1.766	.255	-17.783	4	.000	-19.35600	1.08845	22.37803	16.33397
	<i>Equal variances not assumed</i>			-17.783	2.998	.000	-19.35600	1.08845	22.82098	15.89102

Table 7. Statistical analysis (Independent Samples Test) for % CO<sub>2</sub>

Barrier_film		N	Mean	Std. Deviation	Std. Error Mean
<b>Gas_Mix_CO<sub>2</sub>%_Result</b>	<i>Non_Perforated_Film</i>	3	60.5333	3.21921	1.85861
	<i>Perforated_Film</i>	3	1.1667	1.67506	.96710

	<i>Levene's Test for Equality of Variances</i>	<i>t-test for Equality of Means</i>								
		<i>F</i>	<i>Sig.</i>	<i>t</i>	<i>df</i>	<i>Sig. (2-tailed)</i>	<i>Mean Difference</i>	<i>Std. Error Difference</i>	<i>95% Confidence Interval of the Difference</i>	
									<i>Lower</i>	<i>Upper</i>
<b>Gas_Mix_CO<sub>2</sub>%_Result</b>	<i>Equal variances assumed</i>	1.450	.295	28.335	4	.000	59.36667	2.09517	53.54955	65.18378
	<i>Equal variances not assumed</i>			28.335	3.009	.000	59.36667	2.09517	52.71021	66.02313

Table 8. Statistical analysis (Independent Samples Test)for % N<sub>2</sub>

Barrier_film		N	Mean	Std. Deviation	Std. Error Mean
<b>Gas_Mix_N<sub>2</sub>%Result</b>	<i>Non_Perforated_Film</i>	3	37.4667	3.55293	2.05129
	<i>Perforated_Film</i>	3	77.5000	1.45258	.83865

		Levene's Test for Equality of Variances		t-test for Equality of Means						
		F	Sig.	t	df	Sig. (2-tailed)	Mean Difference	Std. Error Difference	95% Confidence Interval of the Difference	
									Lower	Upper
<b>Gas_Mix_N<sub>2</sub>%_Result</b>	<i>Equal variances assumed</i>	3.583	.131	-18.065	4	.000	-40.03333	2.21610	46.18622	33.88045
	<i>Equal variances not assumed</i>			-18.065	2.650	.001	-40.03333	2.21610	47.64253	32.42413

df= degree of freedom, F= F value, N= number of sample mean used, Sig.= Significance, Std. deviation= Standard deviation, Std. Error Mean= Standard error of the mean, t= Independent t-test

#### 4. CONCLUSIONS

The non-perforated film was able to act as a good barrier film in retaining the initial gas mixture, while the perforated film allowed high % O<sub>2</sub> (between 20.9 to 22.4%) and very low % CO<sub>2</sub> (between 0.15 - 3.1%). The non-perforated film allowed low O<sub>2</sub>% ( 0.63 - 3.91%), which is desirable for food products (Air products, 1995) and high % CO<sub>2</sub> (57 - 66.1%).

Initially, it was hypothesized that the non-perforated film would preserve the physicochemical qualities of the camembert cheese better than the perforated film .

However, the final study revealed that the high % CO<sub>2</sub> may not be appropriate for mould-ripened cheese, such as Camembert. The high % CO<sub>2</sub> under the non-perforated film affected the mould used for ripening. At some point during the study few colonies typical of the *Penicillium camemberti* were observed, whereas high yeast and bacterial counts were noticed for the non-perforated film. The suppression of the mould growth may have given rise to successful competition by anaerobic or microaerophilic yeasts and bacteria.

Therefore, it could be concluded from this study that the use of the perforated film has a good potential for extending the shelf-life of the Camembert cheese.

The result of the study is consistent with the reasoning of some specialists on MAP that mould-ripened cheeses should not be preserved using modified atmosphere (Air products, 1995), because the CO<sub>2</sub> may affect the mould growth and hence the quality of the cheese.

The research revealed that low level CO<sub>2</sub> in the range of 0.2 -4% may be desirable for mould-ripened soft cheeses, such as Camembert.

To the best of our knowledge, this is the first study reporting on the use of MAP

and perforated film for extension of the shelf-life of Camembert cheese. It must, however, be noted that these recommendations on MAP packaging conditions for Camembert cheese correspond to samples from one food company and thus their general application is yet to be verified.

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